

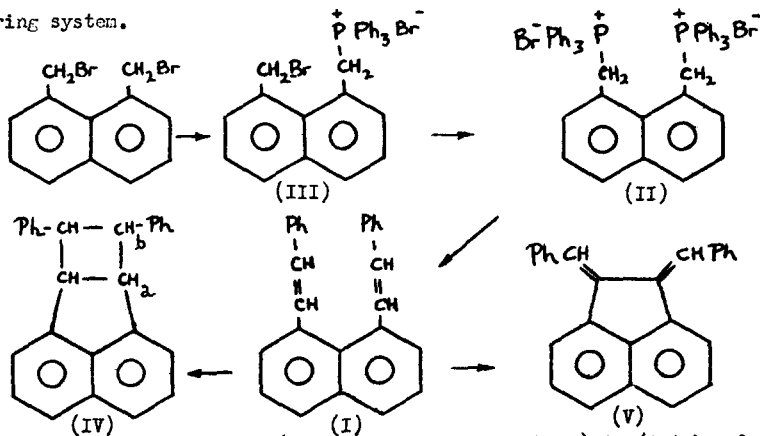
A NEW RING SYSTEM FROM 1,8-DISTYRYLNAPHTHALENE

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We have prepared 1,8-distyrylnaphthalene (I) by a similar method to Bergmann and Agrnant¹ and have subsequently photolysed and dehydrogenated (I) to the acenaphthene derivatives (IV) and (V) respectively; (IV) represents a new ring system.



In the synthesis of (1,8-naphthylenedimethylene)bis-(triphenylphosphonium bromide)(II) we first prepared the monophosphonium bromide (III) by dissolving equimolar amounts of 1,8-naphthalenedimethylbromide² and triphenylphosphine in benzene, and leaving for 24 hours; 1-(3-bromomethyl)naphthylmethyltriphenylphosphonium bromide precipitated in 90% yield: m.p.195°. Calculated (for $\text{C}_{20}\text{H}_{22}\text{Br}_2\text{P}$), Br 27.8%; found Br 27.0%. Equimolar amounts of (III) and triphenylphosphine were dissolved in dimethylformamide and left for 24 hours; the known diphosphonium salt (II) precipitated out of solution.

Treatment of (II) with benzaldehyde in sodium ethoxide, gave 1,8-distyrylnaphthalene: m.p. (from ethanol) 135° ; yield 50%.
 $\lambda_{\text{max}}^{\text{EtOH}}$: 340 m μ ($\log \epsilon = 4.30$), 258 m μ (4.46); τ 2.0-3.2; ν^{KBr} 60, 745, 690 cm^{-1} . Fluorescence spectrum: 427 m μ (broad band). The physical properties of (I) are not exactly in concordance with those described by Bergmann and Agrnant. We were unable to obtain a m.p. above 135° .

Irradiation of (I) in boiling cyclohexane with 2537 \AA light gave cubes of acenaphthene [1,2-a] 3,4-diphenylcyclobutene (IV): m.p. (from ethyl acetate) 335° ; yield 90%. Calculated for $\text{C}_{26}\text{H}_{20}$, C 94.0, H 5.6%; found, C 93.6, H 5.9%. τ_{CH_a} 2.0-2.8, τ_{CH_b} 4.0, τ_{CH_c} 5.2 (ν_{ab} 5 cps.). $\lambda_{\text{max}}^{\text{EtOH}}$ 300 m μ , 225 m μ ($\log \epsilon = 4.9$).

Sublimation of (I) from 5% palladium on charcoal over N_2 at 300° gave a red compound, m.p. 130° (ethanol, 20% yield) confirmed to be 1,2-dibenzylideneacenaphthene³(V) having ultraviolet and infrared spectra identical with an authentic sample.

References

1. E.D. Bergmann and I. Agrnant, J.Org.Chem., 1966, 31, 2407.
2. E.D. Bergmann and T. Szruszkoviz, J.Amer.Chem.Soc., 1953, 75, 2760
3. N. Maxim, Bull.Soc.Chim., 1929, 45, 1137.